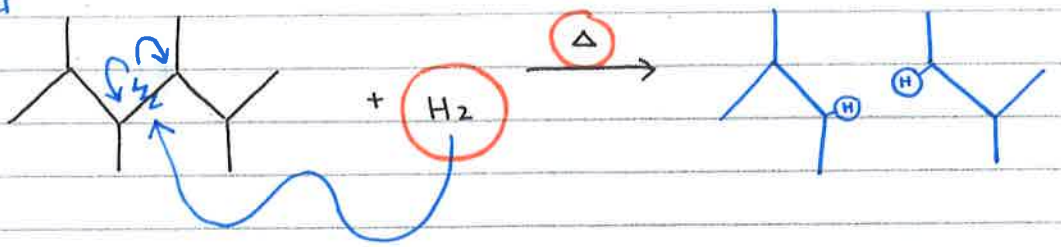
☺ KEY ☺

Friday Study Group
Practice Work

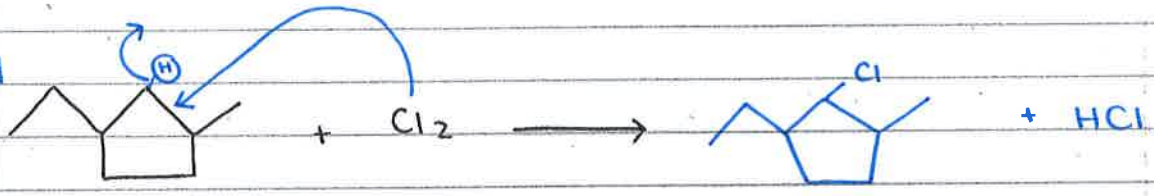
* saturated
* adding H_2



CRACKING

* isomers

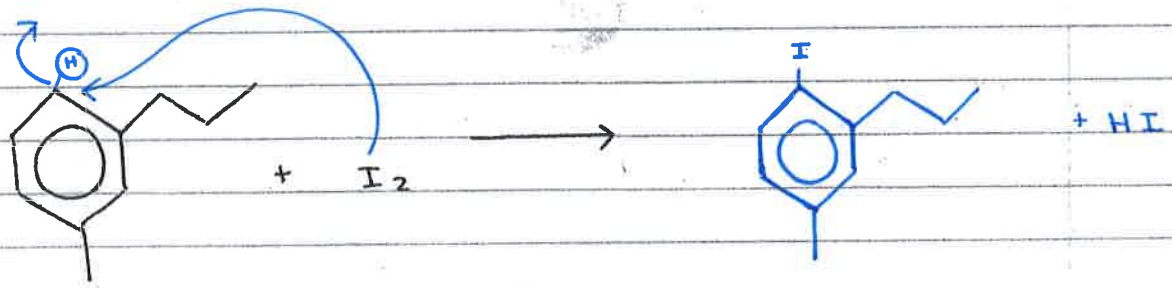
* saturated w/ halogen



SUBSTITUTION

* 2 products
* isomers

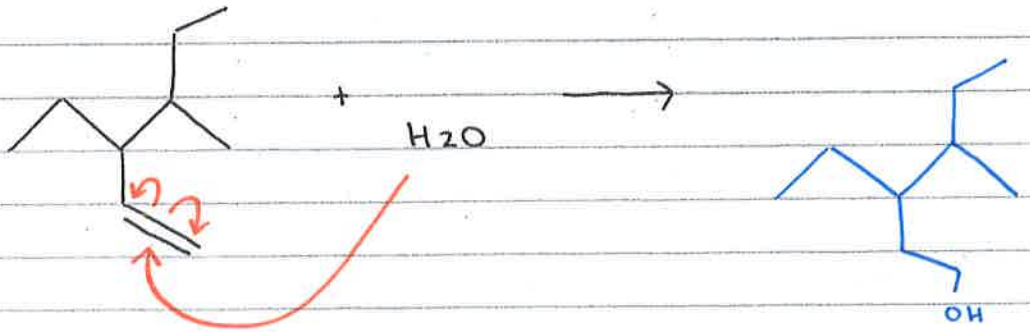
* benzene acts like its saturated



SUBSTITUTION

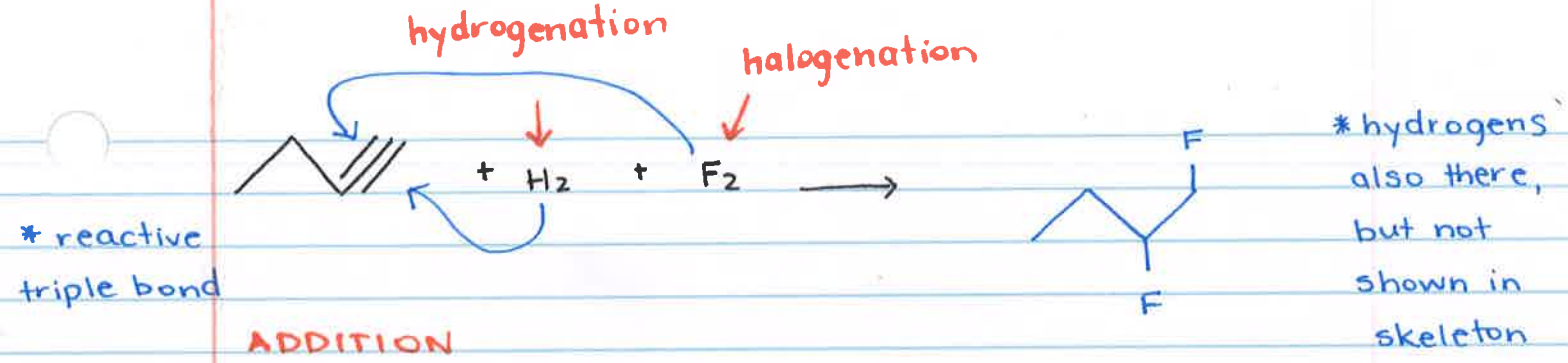
* isomers

* double bond is very reactive

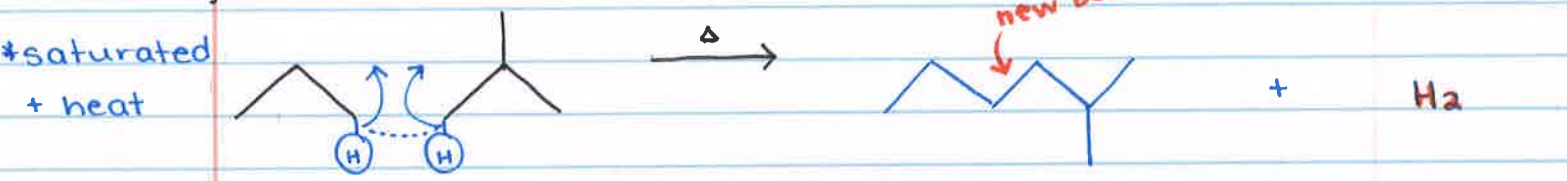


ADDITION

* isomer

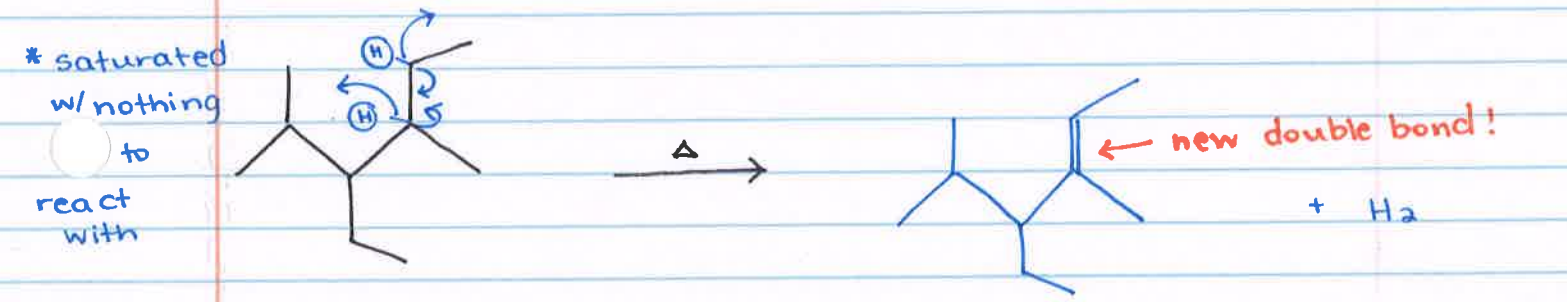


ADDITION



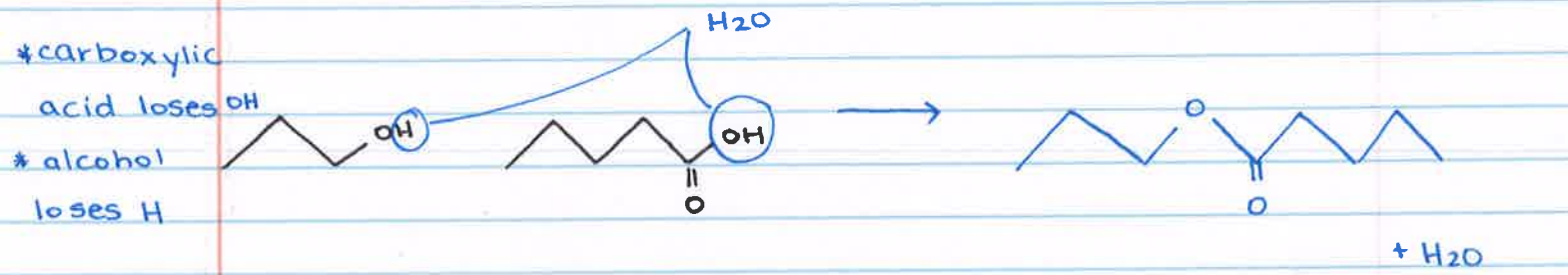
REFORMATION

* isomers

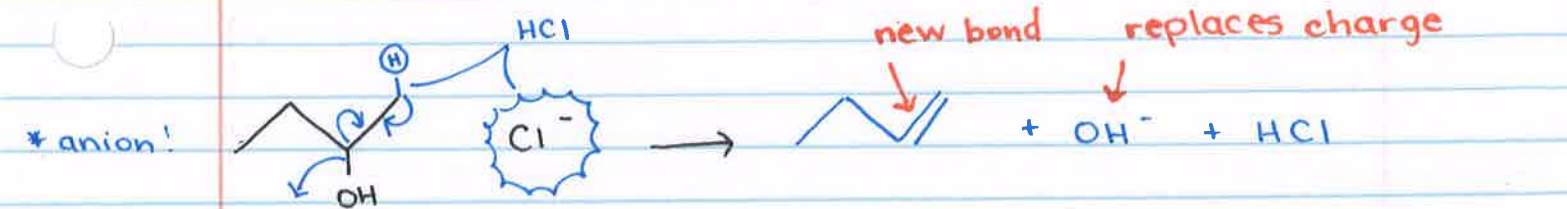


DEHYDROGENATION

* isomers

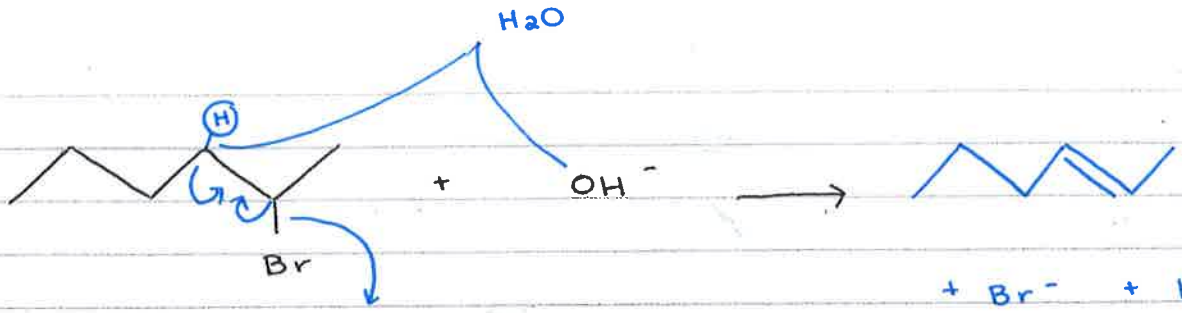


ESTERIFICATION



ELIMINATION

* isomer

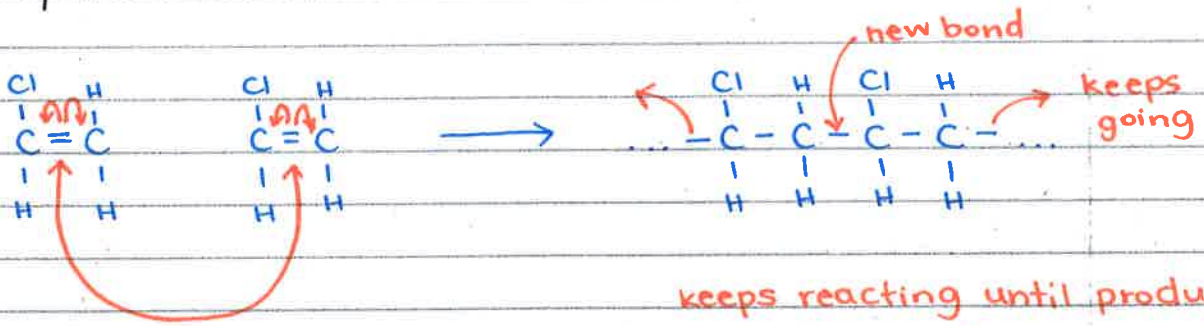


* anion

* isomers

ELIMINATION

Polymerization of chloroethene



* same monomer HOMO

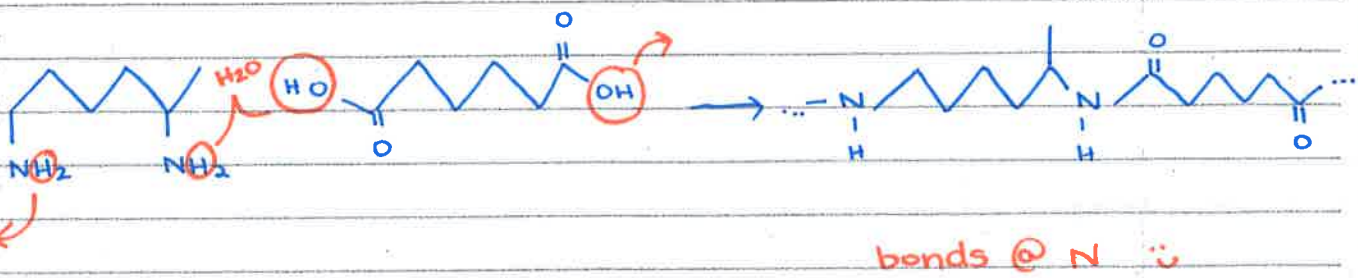
*double bond

ADDITION

keeps reacting until product maxes out

ADDITION POLYMERIZATION (HOMO) (FIBRE)

Polymerization of 1,5-diaminohexane + hexan-1,6 dioic acid



* different monomers COPOLY

bonds @ N :)

* N = protein makes H2O = condensation

CONDENSATION (CO) (PROTEIN)