

Electromagnetic Spectrum Worksheet #1

- In each of the following pairs, circle the form of radiation with the LONGER WAVELENGTH:
 - red light or blue light
 - microwaves or radiowaves
 - infrared radiation or red light
 - gamma rays or UV radiation
- In each of the following pairs, circle the form of radiation with the GREATER FREQUENCY:
 - yellow light or green light
 - x-rays or gamma rays
 - UV radiation or violet light
 - AM radio waves or FM radio waves
- In each of the following pairs, circle the form of radiation with the LOWER ENERGY:
 - red light or blue light
 - microwaves or radiowaves
 - infrared radiation or red light
 - gamma rays or UV radiation
 - yellow light or green light
 - x-rays or gamma rays
 - UV radiation or violet light
 - AM radio waves or FM radio waves
- Springfield's "Classic Rock" radio station broadcasts at a frequency of 102.1 MHz. What is the length of the radio wave in meters?

$$c = f \lambda$$

$$\lambda = c / f$$

$$\lambda = (3.0 \times 10^8 \text{m/s}) / (102.1 \times 10^6 \text{Hz})$$

$$\lambda = 2.938 \text{m}$$

KEY 2016 extra practice

5. A beam of light has a wavelength of 506 nanometers. What is the frequency of the light?
What color is the light?

$$c = f \lambda$$

$$f = c / \lambda$$

$$f = (3.0 \times 10^8 \text{m/s}) / (506 \times 10^{-9})$$

$$f = 5.9289 \times 10^{14} \text{ Hz}$$

6. Blue light has a frequency of 6.98×10^{14} Hertz. Calculate the wavelength of blue light in nanometers.

$$c = f \lambda$$

$$\lambda = c / f$$

$$\lambda = (3.0 \times 10^8 \text{m/s}) / (6.98 \times 10^{14} \text{Hz})$$

$$\lambda = 4.29799 \times 10^{-7} \text{m} = 429.799 \text{ nm}$$